

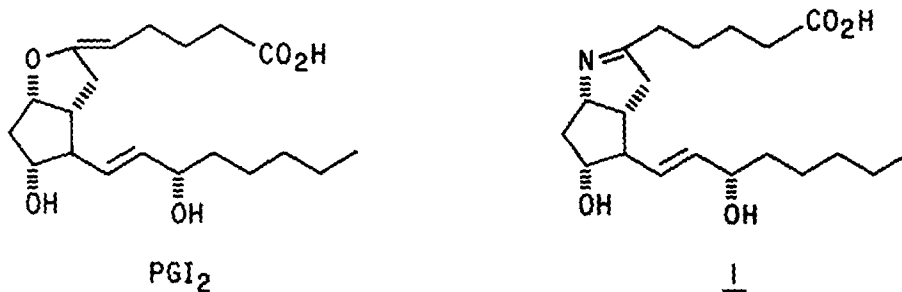
THE SYNTHESIS OF NITROGEN-CONTAINING PROSTACYCLIN ANALOGS

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Prostacyclin^{1,2} (formerly called PGX, now PGI₂) constitutes the most recently discovered addition to the family of biologically potent, unstable metabolites of arachidonic acid formed via the endoperoxide PGH₂. The predominant metabolite of PGH₂ in blood vessel walls, prostacyclin is more potent than PGE₁ or PGD₂ as an inhibitor of platelet aggregation. However, biological evaluation of compounds like the endoperoxides or prostacyclin is complicated by the relative instability of these substances in aqueous solution. In the case of PGH₂, this problem has been partially circumvented by the use of chemically more stable analogs such as cyclic ethers³, azo analogs⁴ and disulfides⁵, which exhibit biological effects similar in many respects to PGH₂ itself, but which cannot be converted enzymatically into the usual array of PGH₂ metabolites. Likewise, we felt that studies currently underway to elucidate the physiological role of prostacyclin and its relationship to other arachidonic acid metabolites would be aided by the availability of prostacyclin analogs which closely resembled PGI₂ in shape, polarity and, hopefully, biological profile, but which possessed greater chemical and metabolic stability.^{1,4}

In this report, we describe the synthesis of 9-deoxy-9 α ,6-nitrilo-PGF_{1 α} 1 and the corresponding cyclic amines in which the oxygen atom of prostacyclin has been replaced by nitrogen.



Nitrilo analog 1 should be similar in polarity to PGI₂, and its sp₂ hybridization at C-6 should allow it to adopt similar conformations as well.

As outlined in Figure 1, PGF_{2 α} methyl ester, 11,15-bis(tetrahydropyranyl ether) 2 was first converted to the 9 β -epimer 4 with triphenylphosphine, benzoic acid and diethyl azodicarboxylate in tetrahydrofuran^{6,7}, followed by treatment of the resulting 9 β -benzoate⁸ 3 with methanolic sodium methoxide. The overall yield of this two-step inversion sequence was 70%. Olefins resulting from elimination at C-9 invariably accompanied the desired product to the extent of 10-20%. This by-product could be minimized by adding the diethyl azodicarboxylate as rapidly as

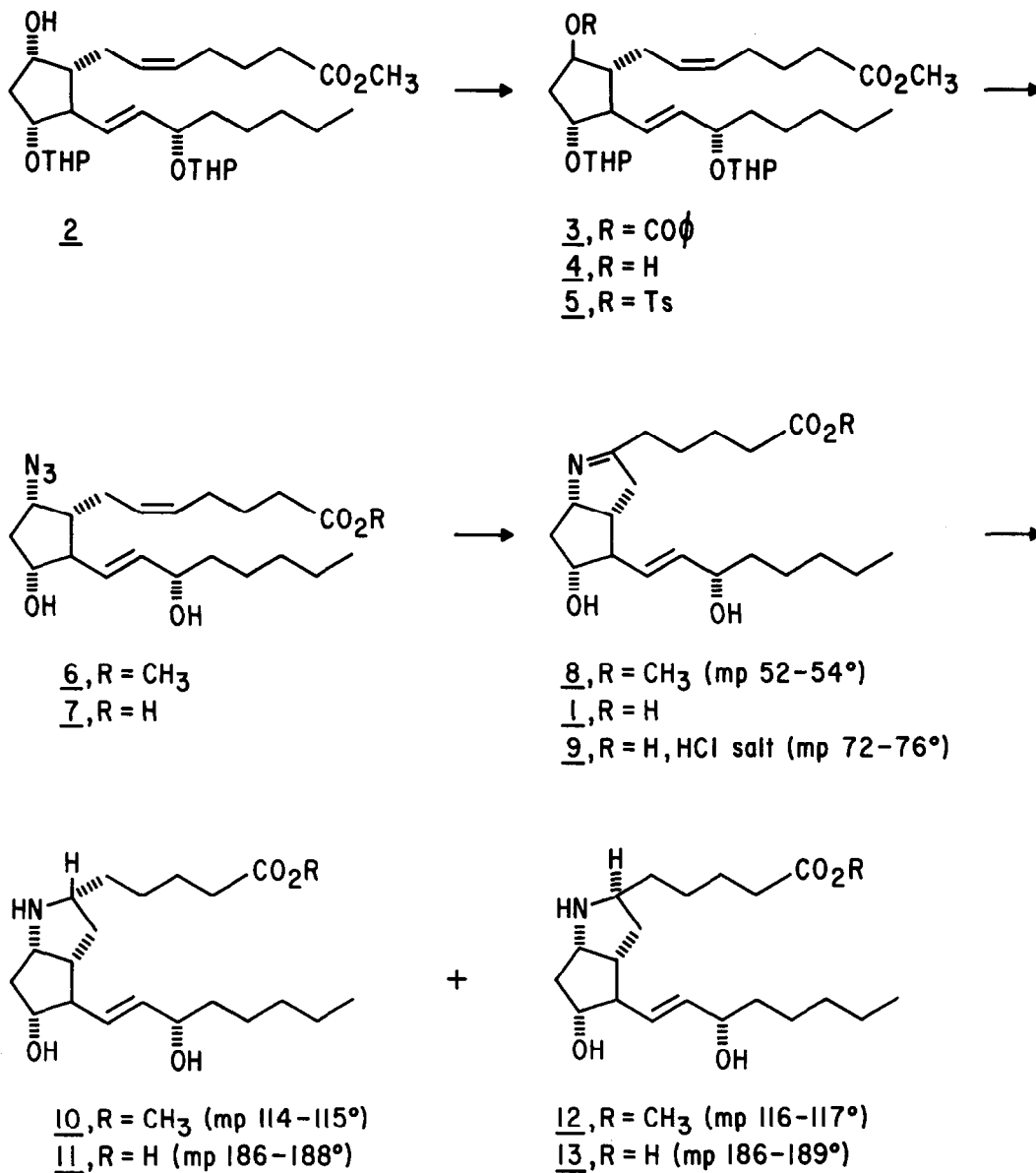


FIGURE 1

physically possible to a cooled, stirred solution of the other reactants. Tosylation of the 9 β -hydroxyl (toluenesulfonyl chloride, pyridine, 18 hr, 25°) and tetrahydropyranyl ether hydrolysis (acetic acid, water, tetrahydrofuran, 40°, 3 hr) afforded the C-9 monotosylate of PGF₂B in 60% yield. Displacement of the 9 β -tosylate with sodium azide in hexamethylphosphoramide (40°, 2 hr) yielded 9 α -azide 6⁸ (essentially quantitative), which upon heating at 70–80°C for 16 hr, either neat or in ethyl acetate, dimethylformamide or dimethoxyethane solution gave crystalline cyclic imine 8 (80% yield over two steps).⁹

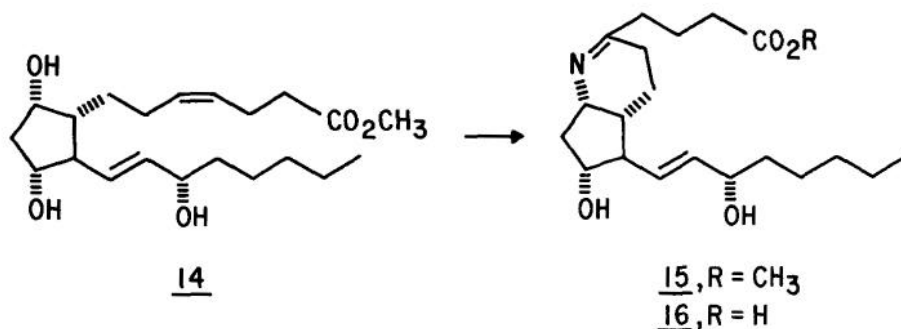
This cycloaddition reaction also proceeded at room temperature or below, but so slowly (~5% in 3 days at 25°) that it was not preparatively useful. Even at the higher temperature, the cycloaddition/N₂ elimination was very clean. No product other than imine 8 was evident on thin layer chromatographic examination of the crude reaction mixture.

The corresponding acid 1 could be obtained either by hydrolysis of ester 8 (followed by XAD-2 resin purification of water-soluble 1), or preferably by hydrolysis of azide ester 6. The acid 7, easily purified and not water soluble, was then subjected to thermal cyclization, yielding imine acid 1 – an amorphous, water-soluble semi-solid, zwitterionic to the extent of about 30%. The 1640 cm⁻¹ infrared band observed in acid 1 and ester 8 is typical of 2-alkyl-1-pyrrolines.¹⁰ The slightly hygroscopic hydrochloride salt 9 formed readily, and in fact both imines 8 and 1 were stable in 1M hydrochloric acid for 3 hr at 25°. (After 24 hr, about 15% decomposition to less polar products.)

Mild acylation of nitrilo ester 8 (acetic anhydride, pyridine, 25°, 18 hr), followed by acid hydrolysis (0.05M HCl, 4:1 THF/H₂O, 1 hr, 25°) afforded 9 α -acetamido-11 α ,15S-diacetoxy-6-oxo-prost-13Z-enoic acid (mp 107.8–108.5°, M⁺ obs. 509.2982; theor. for C₂₇H₄₃NO₈ 509.2988) in excellent yield.

Reduction of imine ester 8 yielded amino esters 10 and 12, which could be purified chromatographically (silica gel; 85 chloroform, 15 methanol, 2 triethylamine; 10 r_f 0.40; 12 r_f 0.34). Each isomer was then hydrolyzed to the corresponding crystalline amino acid (11 and 13 respectively, both completely zwitterionic) and the latter were purified on an XAD-2 resin column (elution with water to remove salts, then methanol to elute the amino acid). The stereochemical assignments at C-6 in amines 10 and 12 were made by comparison of their ¹³C nmr spectra with those of the two isomers of 5,6-dihydro-PGI₂ (PGI₁), which in turn have been identified unequivocally by chemical means.¹¹ In both the oxygen and nitrogen cases, the C-6 and C-9 signals of the more polar (6S) isomer are shifted upfield relative to the less polar (6R) isomer.

Using a reaction sequence similar to that described above for the synthesis of cyclic imine 1, cis- Δ^4 -PGF₁ α methyl ester 14¹² could be converted in six steps⁸ to the 9 α ,5-nitrilo analogs 15 and 16. In this case, the C-11 and C-15 hydroxyls were protected as *t*-butyldimethylsilyl ethers¹³ (5% excess of reagents, 2 hr, 0°, 71% yield) instead of tetrahydropyranyl ethers. Cyclization of the 9 α -azide corresponding to 14 proceeded much more slowly (80°, 44 hr) and in lower yield (44%) here than with the five-membered ring isomer. The C=N absorption at 1660 cm⁻¹ in the infrared spectrum of 15 is consistent with the six membered ring imine structure.¹⁰



Nitrilo analogs 1 and 9 are essentially equivalent to prostacyclin in their ability to inhibit PGH₂-induced or ADP-induced platelet aggregation in human platelet-rich plasma.¹⁵

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